

## **Natural Product Synthesis**

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## **Total Synthesis of Crotophorbolone**

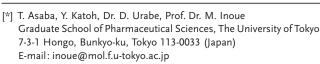
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Abstract: The complex ABC-tricyclic structure of crotophorbolone, a derivative of the tigliane diterpenoids, was assembled by coupling of simple fragments. The six-membered C-ring fragment, having five contiguous stereocenters, was stereoselectively constructed from (R)-carvone. After attachment of the five-membered A-ring through the  $\pi$ -allyl Stille coupling reaction, the α-alkoxy bridgehead radical reaction effected the endo-cyclization of the seven-membered B-ring by forming the sterically congested bond at C9 and C10 stereospecifically and stereoselectively, respectively. Finally, the functional groups on the 5/7/6-membered ring system were manipulated by rhodium-catalyzed C2 olefin isomerization, C13 decarboxylative oxidation, and C4 hydroxylation, thus completing the first total synthesis of crotophorbolone.

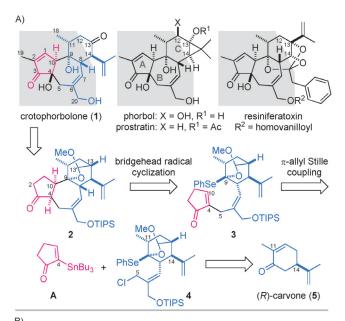
More than one hundred tigliane and daphnane diterpenoids have been identified from the plant families of Euphorbiaceae and Thymelaeaceae, [1] and have been shown to exhibit a remarkably broad range of biological activities. For instance, the tiglianes diacylated phorbol<sup>[2]</sup> and prostratin<sup>[3]</sup> show tumor-promoting and latent HIV-activating activities, [4] respectively, while the daphnane resiniferatoxin [5] possesses an analgesic effect (Scheme 1 A). These natural products share a highly oxygenated 5/7/6-fused ABC-ring system, and only differ in the substitution patterns at the C12,C13,C14 positions. As the C1-C11 substructure highlighted in gray is identical among these compounds, the C12,C13,C14 functionalities are likely to function as important structural elements for diversifying their biological functions.

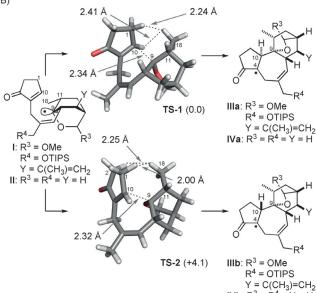
Crotophorbolone (1) was isolated as a degraded compound of phorbol in 1934, and its structure was determined in 1969. [6] In 2010. 1 was identified from the dried plant roots of Euphorbia fischeriana Steud, which has been used as a traditional Chinese medicine for the treatment of edema, ascites, and cancer.[7] The structure of 1 is the same as those of prostratin and resiniferatoxin except for the C13,C14 substitutions. Recently, Wender and co-workers demonstrated efficient conversion of 1 into prostratin through a four-step manipulation of these two stereocenters.<sup>[4]</sup>

Tiglianes, daphnanes, and their derivatives, collectively, have attracted a great deal of attention from the synthetic community for many decades because of their significant biological activities and architecturally complex structures.<sup>[8]</sup> However, even to date, only the groups of Wender<sup>[9]</sup> and



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Scheme 1. Retrosynthetic strategy for crotophorbolone (1). A) Synthetic plan of 1. B) Calculated transition states of the cyclization at UM06-2X/6-31G(d) level. Values in parentheses are relative free energies ( $\Delta G$ in kcal mol<sup>-1</sup>, 1 atm, 298 K).

Cha[10] have reported successful total syntheses of this class of molecules. In this context, we became interested in establishing a new and efficient strategy for assembly of the C1-C12 motif which is common to these natural products. Development of such a strategy would provide access to a number of structurally related targets by minimum modifications of



substrates and reaction sequences. As the initial phase of this campaign, we selected crotophorbolone (1) as a target molecule. Herein we report the total synthesis of 1 from (R)-carvone (5) by employing a  $\pi$ -allyl Stille coupling reaction and an α-alkoxy bridgehead radical mediated cyclization for stereoselective construction of the common C1–C12 tricycle.

The three rings of 1, having five various oxygen-based functionalities, present a daunting synthetic challenge (Scheme 1 A). Crucial to the success of the total synthesis would be efficient formation of the sterically congested C9-C10 bond between the A- and C-rings. To realize this connection, we decided to incorporate the intramolecular α-alkoxy bridgehead radical reaction of 3, which was designed to possess the cyclopentenone and O,Se-acetal moieties as the radical acceptor and donor, respectively.[11] The stereochemically defined and highly reactive bridgehead radical generated from the oxabicyclo[2.2.2]octane 3 would undergo a 7-endo cyclization into 2 with a C9 stereospecific connection of the hindered bond. Appropriate C2 and C4 functionalizations of the A-ring and oxidative cleavage of the C13-C13' bond would transform 2 into the target molecule 1. Accordingly, this radical-based strategy would simplify the stereochemical issues, because only two stereocenters (C4,C10) out of six need to be introduced upon conversion from 3 into 1. The skipped diene structure of the key intermediate 3 would be synthesized from the achiral A-ring A and the chiral C-ring 4 by applying the  $\pi$ -allyl Stille coupling reaction. Pattern recognition of the β-oriented isopropenyl cyclohexane substructure within 4 led us to utilize (R)-carvone (5) as the starting material.[12]

In this approach, the C10 stereocenter must be established upon cyclization of the radical I (Scheme 1B). Since the five chiral centers are concentrated on the C-ring moiety, the stereochemical information of the C-ring of I must be transmitted to the achiral A-ring to produce the correct isomer **IIIa** instead of **IIIb**. Computational modeling studies of II, an abbreviated structure of I, were thus performed to predict the C10 stereoselectivity. Specifically, the two transition-states. TS-1 and TS-2, from II were simulated at the UM06-2X/6-31G(d) level of theory. [13] TS-1, which leads to the desired IVa, was found to be more stable than TS-2 by 4.1 kcal mol<sup>-1</sup>. Close inspection of the three-dimensional structures of TS-1 and TS-2 suggested that the equatorially oriented C11 methyl group influenced their energy difference. While the distances from H18 of the C11 methyl group of **TS-2** to H1 (2.00 Å) and H2 (2.25 Å) are both within the sum of the van der Waals radii (2.40 Å), [14] **TS-1** possesses only one such close contact [H18-1 (2.24 Å), H18-10 (2.41 Å)]. On the basis of the DFT calculations, we expected that the C10 stereocenter would be controlled by the remote C11 stereocenter in a 1,9-relationship.

The synthesis commenced with conversion of (R)-carvone (5) into 11 by introduction of the four new stereocenters (C8, C9, C11, and C13; Scheme 2). Treatment of 5 with TMSCl and MeMgBr in the presence of catalytic FeCl<sub>3</sub> regioselectively produced the dienoxysilane 6.[15] Subsequent vinylogous Mukaiyama aldol reaction of 6 by the action of BF<sub>3</sub>·OEt<sub>2</sub> and CH(OMe)<sub>3</sub><sup>[16]</sup> attached the dimethyl acetal group at the C13 position in an anti-selective manner to the bulky C14 isopropenyl group, leading to 7 as the major product (d.r. = 5:1). The kinetic enolate generated from 7 by LiN $(iPr)_2$ underwent the second aldol reaction with formaldehyde, which was released in situ from **B**.<sup>[17]</sup> The exclusive stereoselectivity in forming the C8 center of 8 was again controlled by the C14 substituent. After TIPS-protection of the primary hydroxy group of 8, the conjugated C11–C12 double bond was reduced under Birch conditions to produce the ketone 10 and alcohol 10' with the thermodynamically stable equatorial C11 methyl group. The over-reduced 10' was in turn oxidized to 10 by TPAP.[18] Then, the sterically demanding lithiated vinyl ether C added to the C9 ketone of 10 equatorially, furnishing the pentasubstituted cyclohexane 11.

Next, the caged C9,C13' bis(acetal) structure 16 was built from 11 through a five-step functional-group manipulation. Upon activation with CSA, the axial C9 alcohol of 11 participated in acetal exchange with the C13' dimethyl acetal to provide the oxabicyclo[2.2.2] octane 12 (d.r. = 1:1 at C13'). The C9 O,Se-acetal was then constructed by conversion of the C9 vinyl ether of 12.[8j] Thus, the vinyl ether 12 was chemoselectively oxidized by mCPBA to the carboxylic acid 13 via the intermediacy of the hydroxy ketone. After mesylation of 13, the mesyloxycarbonyl group of 14 was converted into the PhSe group in a one-pot by Barton ester formation<sup>[19]</sup> and subsequent photoirradiation in the presence of (PhSe)<sub>2</sub>.<sup>[20]</sup> Removal of the TIPS group of **15**, followed by chromatographic separation, gave rise to (13'R)-16 and (13'S)-**16**.

The key intermediate 3 was prepared from (13'R)-16<sup>[21]</sup> by stereoselective construction of the skipped diene structure. The three-carbon extension from the aldehyde 17, which was obtained by SO<sub>3</sub>·pyridine oxidation, was realized by nucleophilic addition of the vinyl lithium E, resulting in formation of 18. The two hydroxy groups of 18 were simultaneously acetylated to afford 19. A reagent combination of Pd<sup>0</sup> and KOAc isomerized the disubstituted olefin 19 into the trisubstituted olefin 20 by  $\pi$ -allyl formation and site-selective addition of acetate to the less-congested primary position.<sup>[22]</sup> After saponification of 20, the more exposed C20 hydroxy group of the resultant diol 21 was regioselectively capped with a bulky TIPS group, and the remaining C5 hydroxy group of 22 was chlorinated to give 4. These two transformations enabled selective activation of C5 over C20 for the  $\pi$ -allyl Stille coupling reaction.<sup>[23]</sup> However, the resulting C6-C7 geometry<sup>[24]</sup> of 4 was partially isomerized under the standard thermal conditions for coupling with A.[25] After screening of additives to avoid this unwanted reaction, CuTC<sup>[26]</sup> was found to significantly accelerate and selectively promote the  $\pi$ -allyl Stille coupling reaction. When 4 was treated with A, CuTC, and K<sub>2</sub>CO<sub>3</sub> in the presence of catalytic [Pd(PPh<sub>3</sub>)<sub>4</sub>] in DMF, the C-C bond formation proceeded, even at 0°C, to afford 3 with no geometrical alteration. Hence, the requisite cis relationship between the radical donor (C-ring) and acceptor (Aring) was secured at this stage.

Most importantly, the O,Se-acetal structure tolerated various nucleophilic and transition-metal reagents from 15 to 3, yet functioned as the radical-generating functional group. Treatment of 3 with (TMS)<sub>3</sub>SiH and V-40 (F) in



Scheme 2. Total synthesis of crotophorbolone (1). Reagents and conditions: a) FeCl<sub>3</sub>, MeMgBr, TMSCl, Et<sub>3</sub>N, N,N'-dimethylpropyleneurea, THF; b) CH(OMe)<sub>3</sub>, BF<sub>3</sub>·OEt<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -50°C, 45%, (d.r. = 5:1, 2 steps); c) LiN(*i*Pr)<sub>2</sub>, THF, -78°C; **B**, 44%; d) TIPSCl, imidazole, DMF; e) Li, NH<sub>3</sub>, THF, -78 °C, 30% for 10′, 36% for 10 (2 steps); f) TPAP, N-methylmorpholine N-oxide, 4 Å MS, CH<sub>2</sub>Cl<sub>2</sub>, 81%; g) C, THF, -78 °C, 89%; h) CSA, benzene, CH(OMe)<sub>3</sub>, 80°C, 96% (d.r. = 1:1); i) mCPBA, pH 7 buffer, CH<sub>3</sub>CN, 10°C, 55%; j) MeSO<sub>2</sub>Cl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0°C; k) **D**, DMAP, toluene; hν, (PhSe)<sub>2</sub>; I) TBAF, CH<sub>3</sub>CN, 60°C, 54% [27% for (13'R)-16, 27% for (13'S)-16, 3 steps]; m) SO<sub>3</sub> pyridine, Et<sub>3</sub>N, DMSO, CH<sub>2</sub>Cl<sub>2</sub>, 80%; n) E, Et<sub>2</sub>O, -78°C; o) Ac<sub>2</sub>O, DMAP, pyridine, 40°C; p) [Pd(PPh<sub>3</sub>)<sub>4</sub>], KOAc, 18-crown-6, THF, 65°C, 56% (3 steps); q) K<sub>2</sub>CO<sub>3</sub>, MeOH, 71%; r) TIPSCI, imidazole, CH<sub>2</sub>Cl<sub>2</sub>, 72%; s) O=C(CCl<sub>3</sub>)<sub>2</sub>, PPh<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>; t) A, [Pd(PPh<sub>3</sub>)<sub>4</sub>], CuTC, K<sub>2</sub>CO<sub>3</sub>, DMF, 0°C, 60% (2 steps); u) (TMS)<sub>3</sub>SiH, F, toluene, 110°C, 69%; v) LiN(iPr)<sub>2</sub>, THF, -78°C; TMSCl, -78°C to 0°C; w) [CH<sub>2</sub>=NMe<sub>2</sub>]<sup>+</sup>I<sup>-</sup>, CH<sub>2</sub>Cl<sub>2</sub>; SiO<sub>2</sub>, n-hexane/EtOAc (10:1), 63% (2 steps); x) RhCl<sub>3</sub>·nH<sub>2</sub>O, EtOH/pH 7 buffer (5:1), 110°C, 68%; y) HCl aq., 1,4-dioxane, 35°C; z) TIPSCl, imidazole, DMF, 10°C; aa) NaClO<sub>2</sub>, NaH<sub>2</sub>PO<sub>4</sub>, 2-methyl-2-butene, tBuOH, H<sub>2</sub>O; bb) TMSOTf, 2,6-lutidine, CH<sub>2</sub>Cl<sub>2</sub>, -20 °C, 54 % (4 steps); cc) **G**, EDCl-HCl, toluene; hv, O<sub>2</sub>, tBuSH; P(OEt)<sub>3</sub>; dd) TESOTf, 2,6lutidine, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 35% (2 steps); ee) NaN(TMS)<sub>2</sub>, THF, -78°C; H, 43% (brsm 62%); ff) TFA, THF/H<sub>2</sub>O (5:1), 0°C; gg) Dess-Martin reagent, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 80% (2 steps); hh) HCl, MeOH/H<sub>2</sub>O (3:1), 70°C, 83%; ii) NaN(TMS)<sub>2</sub>, THF, -78°C; H, -78°C, 31%. brsm = based on recovered starting material, CSA = 10-camphorsulfonic acid, CuTC = copper(I)-thiophene-2-carboxylate, DMAP = 4-(N,N-dimethylamino)pyri $dine, \ DMF = N, N-dimethyl formamide, \ DMSO = dimethyl sulfoxide, \ EDCI-HCl = 1-ethyl-3-(3-dimethyl aminopropyl) carbodiimide \ hydrochloride,$ mCPBA = m-chloroperbenzoic acid, TBAF = tetra-n-butylammonium fluoride, TES = triethylsilyl, Tf = trifluoromethanesulfonyl, TFA = trifluoroacetic acid, THF = tetrahydrofuran, TIPS = triisopropylsilyl, TMS = trimethylsilyl, TPAP = tetra-n-propylammonium perruthenate.



refluxing toluene indeed afforded the α-alkoxy bridgehead radical I (see Scheme 1B), which stereospecifically reacted in a 7-endo manner according to the electronic guidance of the  $\alpha,\beta$ -unsaturated ketone. As expected from the calculations, the C10 stereocenter was completely controlled to deliver the tricyclic ring system 2 as the sole isomer after the C4 stereoselective hydrogen transfer to intermediate IIIa. Therefore, this single step successfully connected the tetrasubstituted and trisubstituted carbon atoms of 1, and installed the three contiguous C4,C9,C10 stereocenters. Notably, the alternative diastereomers, the 6-exo cyclization adduct or the isomerized product of the skipped olefin were not observed, thus demonstrating the high stereo-, regio-, and chemoselectivities of the radical cyclization.

With the tricyclic framework 2 in hand, the C2, C4, and C13 functionalizations from 2 were explored to complete the total synthesis of 1. The olefinic C2 methyl group of 1 was first constructed in three steps. The ketone 2 was regioselectively transformed into the TMS enol ether 23, which was homologated with Eschenmoser's reagent to provide the enone 24 after β-elimination of the dimethyl amino group with silica gel.<sup>[27]</sup> Isomerization of the exo-olefin of 24 into the more stable endo-olefin of 25 was effectively catalyzed by RhCl<sub>3</sub>.<sup>[28]</sup> Next, the extra carbon C13' was exchanged to the C13 hydroxy group by radical oxidation. Before doing so, the cyclic acetal of 25 was opened by acid hydrolysis, and the partially deprotected C20 hydroxy group was re-capped with a TIPS group to afford 27. Oxidation of the aldehyde 27 to the carboxylic acid 28 and subsequent silylation of the C9 tertiary hydroxy group with TMSOTf gave rise to 29. The latter reaction concomitantly induced the C4 epimerization through TMS enol formation and protonation. Finally, the carboxylic acid 29 was stereoselectively transformed into the C13 secondary alcohol 30 by the following one-pot sequence: a) Barton ester formation with **G** and EDCI·HCl; b) peroxide formation by photoirradiation in the presence of oxygen and tBuSH; and c) reductive alcohol formation by P(OEt)<sub>3</sub>.<sup>[29]</sup> The thus constructed C13 hydroxy group of 30 was protected as the TES ether to afford 31.

The C4 hydroxylation, C13 oxidation and deprotection ultimately furnished the target molecule 1 from 31. Among these requisite transformations, stereoselective implementation of the C4 hydroxy group was most problematic, because the oxidation was required to be performed at the sterically shielded angular position without touching the three potentially oxidizable olefins. In fact, proper choice of the substrates and reagents was crucial to the success.[30] When the tricyclic ketone 31 was subjected to NaN(TMS)2, the resultant sodium enolate reacted with Davis' reagent ( $\mathbf{H}$ )<sup>[31]</sup> to deliver the desired C4 stereoisomer 32 as a single isomer. In contrast, treatment of the tetracyclic ketone 25 with NaN-(TMS)<sub>2</sub> and H resulted in sole formation of the undesired C4 stereoisomer I. These contrasting results would originate from the C9 oxygen-based functional groups of **25** and **31**. As the C4 stereochemical outcomes were consistent in the reduction  $(3\rightarrow 2)$  and oxidation  $(25\rightarrow I)$ , the  $\alpha$ -face of the intermediates appeared to be less shielded. The bulky OTMS group at C9 of 31 would override this intrinsic steric bias by blocking the  $\alpha$ -face, thus allowing installation of the  $\beta$ - oriented C4 hydroxy group. In the subsequent reaction with TFA, the TES group of the trisilylated 32 was chemoselectively removed, and the liberated C13 alcohol of 33 was oxidized using the Dess-Martin reagent to the corresponding ketone of 34. Lastly, deprotection of disilylated 34 using a methanolic solution of hydrochloric acid gave rise to crotophorbolone (1). Analytical data of the fully synthetic 1, including the specific rotation, IR, NMR, and HRMS, matched those of natural 1.

In summary, we achieved the first total synthesis of the highly complex diterpenoid crotophorbolone (1) with the 5/7/ 6-ring system from (R)-carvone (5) in 33 linear steps. Design of the unique oxabicyclo[2.2.2]octane 3, as the key intermediate, enabled us to utilize the stereochemically predestined a-alkoxy bridgehead radical for cyclization of the central seven-membered ring of 2. This transformation proved the power and generality of the bridgehead radical reaction for stereoselective construction of the hindered bond within the densely functionalized natural product. Other notable methodological features of the synthesis include: 1) stereoselective installations of four stereocenters from 5; 2) chemoselective construction of the bicyclic bis(acetal) structure 16 equipped with the PhSe group; 3) regioselective formation of the differentially functionalized C6 trisubstituted olefin 4; 4) efficient coupling of the A-ring A and C-ring 4 by the  $\pi$ -allyl Stille coupling using CuTC; and 5) highly optimized C2, C4, and C13 functionalization protocols for conversion of 2 into 1. Further synthetic studies of biologically active tiglianes and daphnanes, which possess the same C1-C12 moiety as 1, are currently being explored based on the present synthetic strategy.

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<sup>[1]</sup> For recent reviews on tigliane and daphnane diterpenoids, see: a) S.-G. Liao, H.-D. Chen, J.-M. Yue, Chem. Rev. 2009, 109, 1092; b) A. Vasas, J. Hohmann, Chem. Rev. 2014, 114, 8579; c) M. J. Durán-Peña, J. M. B. Ares, I. G. Collado, R. Hernández-Galán, Nat. Prod. Rep. 2014, 31, 940; d) H.-B. Wang, X.-Y. Wang, L.-P. Liu, G.-W. Qin, T.-G. Kang, Chem. Rev. 2015, 115, 2975.

<sup>[2]</sup> W. Hoppe, F. Brandl, I. Strell, M. Röhrl, I. Gassmann, E. Hecker, H. Bartsch, G. Kreibich, C. Szczepanski, Angew. Chem. Int. Ed. Engl. 1967, 6, 809; Angew. Chem. 1967, 79, 824.

<sup>[3]</sup> a) A. R. Cashmore, R. N. Seelye, B. F. Cain, H. Mack, R. Schmidt, E. Hecker, Tetrahedron Lett. 1976, 17, 1737; b) G. A. Miana, M. Bashir, F. J. Evans, Planta Med. 1985, 51, 353.



- [4] a) P. A. Wender, J.-M. Kee, J. M. Warrington, Science 2008, 320, 649; b) E. J. Beans, D. Fournogerakis, C. Gauntlett, L. V. Heumann, R. Kramer, M. D. Marsden, D. Murray, T.-W. Chun, J. A. Zack, P. A. Wender, Proc. Natl. Acad. Sci. USA 2013, 110, 11698.
- [5] M. Hergenhahn, W. Adolf, E. Hecker, Tetrahedron Lett. 1975, 16, 1595.
- [6] a) B. Flaschenträger, F. F. Falkenhausen, Justus Liebigs Ann. Chem. 1934, 514, 252; b) H. W. Thielmann, E. Hecker, Justus Liebigs Ann. Chem. 1969, 728, 158.
- [7] a) H.-B. Wang, W.-J. Chu, Y. Wang, P. Ji, Y.-B. Wang, Q. Yu, G.-W. Qin, J. Asian Nat. Prod. Res. 2010, 12, 1038; 15,16dihydrocrotophorbolone (langduin A) was isolated from the same source; b) Q.-G. Ma, W.-Z. Liu, X.-Y. Wu, T.-X. Zhou, G.-W. Qin, Phytochemistry 1997, 44, 663.
- [8] For recent synthetic studies of tigliane and daphnane diterpenoids, see: a) J. H. Rigby, P. C. Kierkus, D. Head, Tetrahedron Lett. 1989, 30, 5073; b) L. M. Harwood, T. Ishikawa, H. Phillips, D. Watkin, J. Chem. Soc. Chem. Commun. 1991, 527; c) W. G. Dauben, J. Dinges, T. C. Smith, J. Org. Chem. 1993, 58, 7635; d) L. A. Paquette, D. R. Sauer, S. D. Edmondson, D. Friedrich, Tetrahedron 1994, 50, 4071; e) K. Sugita, C. F. Neville, M. Sodeoka, H. Sasai, M. Shibasaki, Tetrahedron Lett. 1995, 36, 1067; f) T. V. Ovaska, S. E. Reisman, M. A. Flynn, Org. Lett. 2001, 3, 115; g) S. R. Jackson, M. G. Johnson, M. Mikami, S. Shiokawa, E. M. Carreira, Angew. Chem. Int. Ed. 2001, 40, 2694; Angew. Chem. 2001, 113, 2766; h) P. C. B. Page, C. M. Hayman, H. L. McFarland, D. J. Willock, N. M. Galea, Synlett 2002, 583; i) C. Stewart, R. McDonald, F. G. West, Org. Lett. 2011, 13, 720; j) K. Murai, S. Katoh, D. Urabe, M. Inoue, Chem. Sci. 2013, 4, 2364; k) A. J. Catino, A. Sherlock, P. Shieh, J. S. Wzorek, D. A. Evans, Org. Lett. 2013, 15, 3330; l) G. Tong, Z. Liu, P. Li, Org. Lett. 2014, 16, 2288; m) A. H. E. Hassan, J. K. Lee, A. N. Pae, S.-J. Min, Y. S. Cho, Org. Lett. 2015, 17, 2672.
- [9] For total synthesis of phorbol, see: a) P. A. Wender, H. Kogen, H. Y. Lee, J. D. Munger, Jr., R. S. Wilhelm, P. D. Williams, J. Am. Chem. Soc. 1989, 111, 8957; b) P. A. Wender, F. E. McDonald, J. Am. Chem. Soc. 1990, 112, 4956; c) P. A. Wender, K. D. Rice, M. E. Schnute, J. Am. Chem. Soc. 1997, 119, 7897; for total synthesis of daphnane diterpenoids resiniferatoxin and yuanhuapin analogues, see: d) P. A. Wender, C. D. Jesudason, H. Nakahira, N. Tamura, A. L. Tebbe, Y. Ueno, J. Am. Chem. Soc. 1997, 119, 12976; e) P. A. Wender, N. Buschmann, N. B. Cardin, L. R. Jones, C. Kan, J.-M. Kee, J. A. Kowalski, K. E. Longcore, Nat. Chem. 2011, 3, 615.
- [10] K. Lee, J. K. Cha, J. Am. Chem. Soc. 2001, 123, 5590.
- [11]  $\alpha$ -Alkoxy bridgehead radical reactions were utilized for the twoand three-component coupling reactions in our laboratory. Its application to the cyclization was not previously realized: a) D. Urabe, H. Yamaguchi, M. Inoue, Org. Lett. 2011, 13, 4778; b) D. Urabe, M. Nagatomo, K. Hagiwara, K. Masuda, M. Inoue, Chem. Sci. 2013, 4, 1615; c) D. Kamimura, D. Urabe, M. Nagatomo, M. Inoue, Org. Lett. 2013, 15, 5122; d) M. Nagatomo, M. Koshimizu, K. Masuda, T. Tabuchi, D. Urabe, M. Inoue, J. Am. Chem. Soc. 2014, 136, 5916. See also Ref. [8j].
- [12] For selected reviews on total synthesis of terpenoids, see: a) K. C. Nicolaou, D. Vourloumis, N. Winssinger, P. S. Baran, Angew. Chem. Int. Ed. 2000, 39, 44; Angew. Chem. 2000, 112, 46; b) R. M. Wilson, S. J. Danishefsky, J. Org. Chem. 2007, 72, 4293; c) T. J. Maimone, P. S. Baran, Nat. Chem. Biol. 2007, 3, 396; d) D. Urabe, T. Asaba, M. Inoue, Chem. Rev. 2015, 115, 9207.
- [13] a) Y. Zhao, D. G. Truhlar, Theor. Chem. Acc. 2008, 120, 215; b) Y. Zhao, D. G. Truhlar, J. Phys. Chem. A 2008, 112, 1095.
- [14] H. Wang, K. Michalak, M. Michalak, G. Jiménez-Osés, J. Wicha, K. N. Houk, J. Org. Chem. 2010, 75, 762.

- [15] a) M. E. Krafft, R. A. Holton, J. Am. Chem. Soc. 1984, 106, 7619; b) O. Takazawa, H. Tamura, K. Kogami, K. Hayashi, Bull. Chem. Soc. Jpn. 1982, 55, 1907.
- [16] For a recent review on the vinylogous Mukaiyama aldol reaction, see: M. Kalesse, M. Cordes, G. Symkenberg, H.-H. Lu, Nat. Prod. Rep. 2014, 31, 563.
- [17] G. Deguest, L. Brishoff, C. Fruit, F. Marsais, Org. Lett. 2007, 9, 1165.
- [18] S. V. Ley, J. Norman, W. P. Griffith, S. P. Marsden, Synthesis 1994 639.
- [19] a) D. H. R. Barton, D. Crich, W. B. Motherwell, J. Chem. Soc. Chem. Commun. 1983, 939; for reviews on the Barton ester, see: b) D. Crich, L. Quintero, Chem. Rev. 1989, 89, 1413; c) M. F. Saraiva, M. R. C. Couri, M. L. Hyaric, M. V. de Almeida, Tetrahedron 2009, 65, 3563.
- [20] D. H. R. Barton, D. Bridon, S. Z. Zard, Tetrahedron Lett. 1984, 25, 5777.
- [21] The diastereomer (13'S)-16 was also transformed into 26 in a similar overall yield to (13'R)-16 by the same 13-step sequence. See the Supporting Information for details.
- [22] A. Stolle, J. Ollivier, P. P. Piras, J. Salaün, A. de Meijere, J. Am. Chem. Soc. 1992, 114, 4051.
- [23] For a review on Stille couplings, see: a) V. Farina, V. Krishnamurthy, W. J. Scott, Org. React. 1997, 50, 1; for recent applications of  $\pi$ -allyl Stille couplings on natural product synthesis, see: b) C. D. Vanderwal, D. A. Vosburg, S. Weiler, E. J. Sorensen, J. Am. Chem. Soc. 2003, 125, 5393; c) S. A. Snyder, E. J. Corey, J. Am. Chem. Soc. 2006, 128, 740; d) W. D. Shipe, E. J. Sorensen, J. Am. Chem. Soc. 2006, 128, 7025.
- [24] For a review on isomerization through interconvertible  $\pi$  and  $\sigma$ allyl metal species, see: a) G. Consiglio, R. M. Waymouth, Chem. Rev. 1989, 89, 257; for configurational stability of the double bond under the  $\pi$ -allyl Stille coupling, see: b) L. D. Valle, J. K. Stille, L. S. Hegedus, J. Org. Chem. 1990, 55, 3019.
- [25] The  $\pi$ -ally Stille coupling of the corresponding allyl carbonate by using catalytic [Pd(PPh<sub>3</sub>)<sub>4</sub>] and stoichiometric amounts of CuCl, LiCl, and Et<sub>3</sub>N in DMSO at 80°C resulted in the formation of a 1:1 mixture of 3 and its regioisomer in 26% combined yield.
- [26] a) G. D. Allred, L. S. Liebeskind, J. Am. Chem. Soc. 1996, 118, 2748; For the use of CuTC in Pd-catalyzed Stille couplings, see: b) D. R. Williams, K. G. Meyer, J. Am. Chem. Soc. 2001, 123, 765; c) H. Fuwa, N. Kainuma, K. Tachibana, M. Sasaki, J. Am. Chem. Soc. 2002, 124, 14983; d) A. Fürstner, C. Nevado, M. Tremblay, C. Chevrier, F. Teplý, C. Aïssa, M. Waser, Angew. Chem. Int. Ed. 2006, 45, 5837; Angew. Chem. 2006, 118, 5969; e) L. M. Bishop, J. E. Barbarow, R. G. Bergman, D. Trauner, Angew. Chem. Int. Ed. 2008, 47, 8100; Angew. Chem. 2008, 120, 8220; f) J. Li, P. Yang, M. Yao, J. Deng, A. Li, J. Am. Chem. Soc. 2014, 136, 16477.
- [27] a) J. Schreiber, H. Maag, N. Hashimoto, A. Eschenmoser, Angew. Chem. Int. Ed. Engl. 1971, 10, 330; Angew. Chem. 1971, 83, 355; b) S. Danishefsky, T. Kitahara, R. McKee, P. F. Schuda, J. Am. Chem. Soc. 1976, 98, 6715.
- [28] a) B. W. Disanayaka, A. C. Weedon, Synthesis 1983, 952; b) A. Nickel, T. Maruyama, H. Tang, P. D. Murphy, B. Greene, N. Yusuff, J. L. Wood, J. Am. Chem. Soc. 2004, 126, 16300.
- [29] D. H. R. Barton, D. Crich, W. B. Motherwell, Tetrahedron 1985, 41. 3901.
- [30] See the Supporting Information for the optimization.
- [31] F. A. Davis, A. C. Sheppard, B.-C. Chen, M. S. Haque, J. Am. Chem. Soc. 1990, 112, 6679.

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